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This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important topics.

Sweet 16 Chemistry Compound Tournament

The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these

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Important topics: The rules for filling out the Tournament brackets are simple: For the first round, the student determines the chemical for-

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Publication No. 10510 With spring just around the corner, your students' thoughts will soon be turning to sunshine, prom, and the NCAA basketball tournament. This clever activity combines the ever-popular basketball office pool with a review of the formulas of ionic compounds and their solubility.

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Many tournaments and field trips have allowed Kate and I to both do some long term substitution. On one of these weeks, I had a lower end third form class (9th grade). We were reading Trifles, a play about a woman who kills her husband in rural Iowa, by Susan Glaspell. I love this play, and wanted to get the students involved too.

Katie In Oz ☐ LiveJournal

The winner of the tournament is the compound in the finals that has the larger molar mass. Two versions of the Sweet 16 Chemistry Compound Tournament are enclosed. One uses traditional chemical names [e.g. ferrous, cupric, etc] and the other uses the Stock system for chemical names

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